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Sequential growth of self-organized epitaxial FeSi $_2$ and CoSi $_2$ nanostructures on Si(111)-7 \times 7 surfaces

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ABSTRACT

Epitaxial γ -FeSi $_2$ and CoSi $_2$ nanostructures, grown on a vicinal Si(111)-7 \times 7 surface by sequentially depositing 1-monolayer (ML) of Fe and 1ML of Co on a hot Si substrate under ultrahigh vacuum condition, have been investigated by in-situ scanning tunneling microscopy, ex-situ field emission scanning electron microscopy and x-ray photoemission spectroscopy (XPS). While γ -FeSi $_2$ nanostructures have grown as nanowires along the three equivalent $\langle 110 \rangle$ directions on the Si(111) surface, CoSi $_2$ nanoislands have grown as equilateral triangular and trapezoidal islands. Such self-organized nanostructures may find applications in nanoscale devices including S-F hybrids for quantum technology, as CoSi $_2$ is a superconductor (S) and γ -FeSi $_2$ is a ferromagnet (F).

1. Introduction

The transition metal silicides, especially chromium disilicide, manganese disilicide, iron disilicide, cobalt disilicide, nickel disilicide etc. are very important active components of nanoelectromechanical devices due to their very special mechanical [1–5], thermal [1,2,4], optical [3,6-10], electrical [3,7] and magnetic [6,11-20] properties. These silicides are very useful in the nanodevices because of their high thermal stability nearly up to $\sim 1000~{\rm ^{\circ}C}$, and they form very sharp interface with the substrate and the Schottky barrier height can be tuned by proper choice of silicide material [1,2,4]. Among these metal silicides, iron disilicide (FeSi2) and cobalt disilicide (CoSi2), are two most important silicides. For nanoelectronics, it is important to understand the silicide/silicon systems in the nanoscale [18-22].

Another potential development would arise from the fact that $CoSi_2$ is a superconductor, in the bulk form <code>[23,24]</code> as well as in epitaxial structures on silicon <code>[25,26]</code>, and $\gamma\text{-FeSi}_2$ is a ferromagnet <code>[27-30]</code>. Bulk $CoSi_2$ has a superconducting transition temperature (TC) of ~ 1.4 K <code>[23]</code>, very similar to that of aluminum (TC = 1.2 K)- a popular material for making superconducting quantum circuits <code>[31]</code>. The intrinsically low

noise properties of superconducting epitaxial CoSi2 thin films on Si, even compared to aluminum, could be utilized for developing quiet qubits and scalable superconducting circuits for future quantum computing [26]. We have earlier grown epitaxial $CoSi_2$ [32-34] and γ -FeSi₂ [35] individually on silicon substrates. However, growing γ-FeSi₂ and CoSi₂ nanostructures together on a single substrate assumes importance in view of the application of ferromagnet-superconductor hybrids (FSH) in quantum technology [36]. The so-called FSHs are spatially separated but closely located ferromagnets and superconductors. In general, many different kinds of hybrid systems are the candidates for the development of quantum technologies. Hybrid quantum systems composed of different physical components with complementary functionalities may provide multi-tasking capabilities [37]. Epitaxial growth of both FeSi₂ and CoSi2 nanostructures on the same silicon substrate may lead to the integration of the relevant quantum technologies with the wellestablished silicon technology.

Numerous research groups have widely investigated the growth of self-organized epitaxial FeSi₂ nanostructures on Si(111)-7 \times 7 surfaces [11,38-44]. Also there are reports on the growth of self-organized epitaxial CoSi₂ on the Si(111)-7 \times 7 surfaces [22,45-50]. But there is

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no report on the epitaxial growth of more than one silicide on the same Si(111)-7 \times 7 surfaces. Also there is no investigation on how Co reacts when it is deposited on the hot Si(111)-7 \times 7 surfaces which are predecorated with FeSi2 nanostructures. In other words, there is no report on the influence of one silicide (here, FeSi2) nanostructures on the growth of another silicide on the same Si(111)-7 \times 7 surfaces. In this article an attempt has been made to address this issue. Growth of more than one silicide may play a significant role if the device fabrication via self-organized growth process can be realized in future. In this paper, for the first time we present the fabrication of concomitant self-organized epitaxial FeSi2 NWs and CoSi2 nanodots or nanoislands on 3°-miscut vicinal Si(111)-7 \times 7 surfaces. The detailed morphology of the FeSi₂ and CoSi₂ nanostructures has been investigated by in-situ scanning tunneling microscopy (STM) measurements and ex-situ field emission scanning electron microscopy (FE-SEM) measurements. The formation of the two distinct disilicides has been confirmed by X-ray photoelectron spectroscopy (XPS) measurements.

2. Experimental

Atomically clean $3^{\circ}\pm0.1^{\circ}$ vicinal Si(111)-7 \times 7 surfaces were prepared by degassing the substrate at \sim 700 °C for 14–16 h and then flashing the sample at \sim 1250 °C for one min under ultrahigh vacuum (UHV) environment (\sim 5 \times 10 $^{-10}$ mbar) in a molecular beam epitaxy (MBE) chamber. We raised the substrate temperature to \sim 1250 °C, kept there for one min, and then cooled it down to \sim 870 °C and kept at that

temperature for 30 min and then allowed the substrate to attain the room temperature slowly by switching off the sample heating direct current. The clean surface thus prepared possesses atomically flat Si (111)-7 \times 7 surfaces with small terraces. One monolayer (ML) (1ML = $9.59 \times 10^{14} \text{ atoms/cm}^2$) of Fe atoms has been deposited on the hot substrate at ~ 600 °C. This method produces γ -FeSi₂ [27,28,35]. After in-situ STM measurement of the Fe-deposited Si(111)-7 \times 7 surfaces, the FeSi₂ decorated surface is transferred to the MBE growth chamber and then heated again to $\sim 600\,^{\circ}\text{C}$ to deposit 1.0 ML Co onto it. As the MBE and the STM chambers are interconnected, both under UHV condition, the sample is never exposed to air. The Fe- and Co-deposited Si(111)-7 \times 7 surface was investigated by in-situ STM. Then the sample was taken out of the MBE-STM chamber to perform FE-SEM and subsequently XPS measurements. The XPS measurements were performed by using PHI 5000 VERSAPROBE II, Physical Electronics system equipped with a monochromatic Aluminum K_{α} (1486.7 eV) high flux focused X-ray source and a multi-channeltron hemispherical electron energy analyzer. All the spectra were collected at an emission angle of 45° with the base vacuum of 4.0×10^{-10} mbar. The binding energy calibration was done by measuring C 1 s keeping at 284.6 eV. The total energy resolution, estimated from the width of the Fermi edge, was about 400 meV for monochromatic Aluminum K_{α} line with pass energy 11.750 eV. A charge neutralizer was utilized in order to compensate the surface charging of the samples.

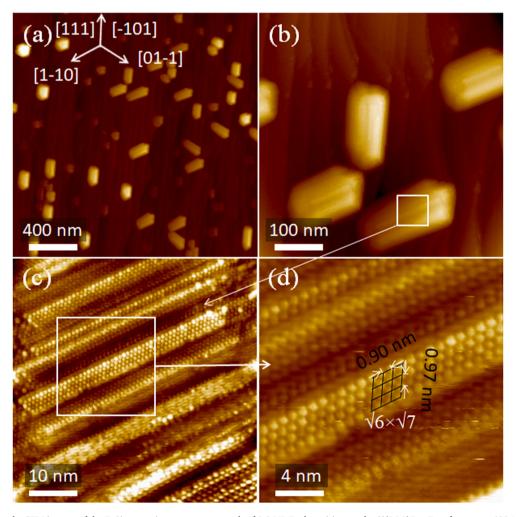


Fig. 1. Fig. (a-d) show the STM images of the $FeSi_2$ nanowires grown as a result of 1.0 ML Fe deposition on the $Si(111)7 \times 7$ surfaces at ~ 600 °C. The filled state ($V_g = -1.6$ V and $I_t = 0.2$ nA) STM images show the atomic arrangement of the top surface of the $FeSi_2$ atomic rows.

3. Results and discussion

3.1. Self-organized epitaxial FeSi $_2$ nanostructures on Si(111)-7 \times 7 surfaces via reaction deposition epitaxy

STM images of an atomically clean 3°-miscut vicinal Si(111) surface, recorded at a bias voltage, $V_g = 1.9 \text{ V}$ and tunneling current, $I_t = 0.2 \text{nA}$, have shown typical (7×7) surface reconstruction on the terraces; the average terrace width of this vicinal clean surface is $\sim 50 \text{ nm}$ (not shown here). Fig. 1 shows STM images of the FeSi2nano-islands grown on the 3°-miscut vicinal Si(111)-7 \times 7 reconstructed surface, recorded at a bias voltage $V_g =$ -1.6 V and tunneling current, $I_t =$ 0.2nA. The heights of the nano-islands lie in the range from 8.5 to 14.0 nm. The FeSi₂nano-islands are elongated along three in-plane (110) crystallographically equivalent directions. The atomic resolution STM image on the top surfaces of these FeSi2nano-islands shows well-ordered atomic arrangements on stripe-like structures (Fig. 1(c, d)). On relatively wider stripes, the unit cell shows a $(\sqrt{6} \times \sqrt{7})$ reconstruction, as shown in Fig. 1(d). Fig. 2(a-d) shows the magnified images of a single FeSi2nanowire. Fig. 2(a) is the STM image recorded at bias voltage, $V_g=1.6\ V$, tunneling current $I_t=$ 0.2nA and Fig. 2 (b) is the current image of the former. From Fig. 2(a, c), the Si(111)7 \times 7 reconstruction is clearly visible surrounding the FeSi2nanowire. This indicates Volmer-Weber growth of FeSi2. The different facets of the pyramid shaped nano-island of FeSi2 possess different surface atomic arrangements as seen from Fig. 2(c-d). The nature of the

FeSi $_2$ nanowires here are different from those on Si(110) surfaces [35]. The FeSi $_2$ nanowires in Ref. 35, possess comb-like structures on the two side walls, which could be related to the (16 \times 2) reconstruction on the Si(110) surface, and all the nanowires grow along one direction because of the twofold symmetry of the Si(110) surface. Here the FeSi $_2$ nanowires grow along the three $\langle 110 \rangle$ in-plane directions, due to the threefold symmetry of the Si(111) surface, and they do not show comb-like features.

3.2. Self-organized epitaxial CoSi $_2$ and FeSi $_2$ nanostructures on Si(111)-7 \times 7 surfaces via reaction deposition epitaxy

Fig. 3 (a, c) show the STM images of the $FeSi_2$ and $CoSi_2$ nanostructures grown on $Si(1\,1\,1)$ -7 \times 7 surfaces. Fig. 3(b, d) show the current images of the corresponding STM images in Fig. 3(a, c). Comparing the morphology of the nanostructures of these Fe- and Co-deposited surface with that of only Fe-deposited surface, it is evident that triangular and trapezoidal nanostructures have appeared (encircled in Fig. 3(a, c)), in addition to those which had already grown on the surface when only Fe was deposited. Current image of Fig. 3(c) in Fig. 3(d) clearly shows the triangular and trapezoidal nanostructures and also the nanowires. The current image (derivative mode) displays enhanced information when the z-scale varies fast. Here, we can see this feature of the sidewall facets of the silicide nanowires and nanoislands more prominent in Fig. 3(b, d). The edges of the triangular and the trapezoidal islands are along the

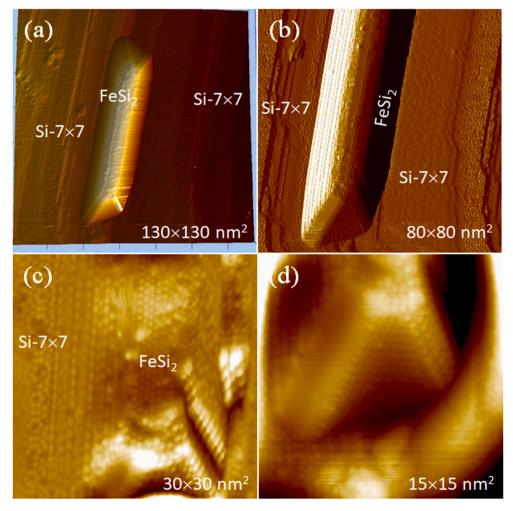


Fig. 2. (a) shows the 3D view of the STM image of a FeSi $_2$ NW grown on Si(111)-7 \times 7 clean surfaces. (b) shows the current image of the STM image of the same FeS $_2$ NW. Surrounding the NW, the 7 \times 7 reconstruction is sustained. (c, d) show the atomically resolved STM images of the NW. Different facets of the NW show different atomic arrangements.

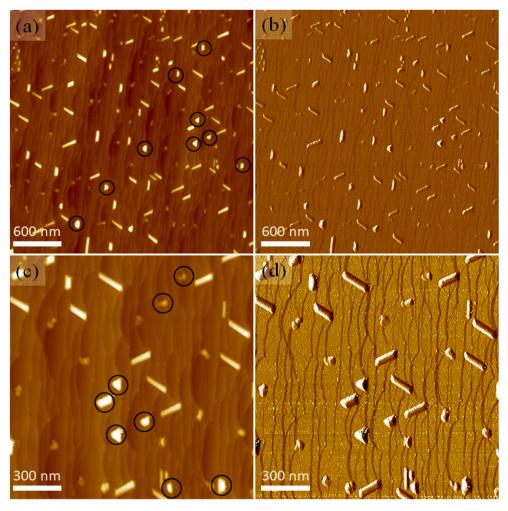


Fig. 3. (a) and (c) show the STM images (bias voltage, $V_g = 1.9 \text{ V}$ and tunneling current, $I_t = 0.2 \text{nA}$) of the FeSi₂ and CoSi₂ NWs and nanostructures grown on the Si (111)-7 \times 7 surface. (b, d) show the current images of the STM images shown in (a) and (c) respectively.

three in-plane $\langle 110 \rangle$ crystallographically equivalent directions. For CoSi₂ growth on Si(111)-7 \times 7 surfaces, these growth features have earlier been observed [32,48]. From Fig. 3(a) we notice that the trapezoidal islands are preferentially aligned along the step edges – a feature observed in other cases of epitaxial island growth on vicinal Si(111) surfaces [50,51].

Fig. 4(a-d) shows the STM image of $FeSi_2$ NWs and $CoSi_2$ nanostructures grown on the vicinal Si(111)-7 \times 7 surface. Fig. 4(b) shows the magnified image of marked part in (a). Fig. 4(c) shows the three-dimensional view of the nanoisland in the upper part seen in (b) and (d) is its current image. The nanoisland possesses five distinct side facets. The high resolution STM image reveals that there are different atomic arrangements on these side facets.

Fig. 5 shows the STM image of a flat terrace silicide island in (a-c). Fig. 5(c) shows the 2×2 reconstructed atomically resolved STM image of the top surface of the silicide island of the box marked location in (b). Fig. 5(d) is the Fast Fourier transform (FFT) image of Fig. 5(c). The unit cell dimension of the FFT image is 1.31 nm $^{-1}$, which corresponds to the direct lattice unit cell of side 0.76 nm.

SEM investigation also supports the STM morphology investigation. Fig. 6 shows the SEM image recorded at $5.0~\rm kV$. In Fig. 6, the FeSi2 nanowires have been marked as 'F' and the CoSi2 triangular islands have been marked as 'C' in the SEM image. Though, the nanowires are mainly made up of FeSi2, it is to be worth mentioned that out of the nanowires as seen in the STM as well as SEM images, there may be a very few CoSi2 nanowires too. As it is reported earlier, deposition of cobalt on the Si

(111)-7 \times 7 results in the formation of triangular as well as a very few cobalt disilicide nanowires with them [45,48]. The line profile plot on the FeSi₂ and CoSi₂-grown sample reveals that the compact nanoislands are taller than the elongated nanowires. The height of the nanodots lies in the range of 25.0–30.0 nm whereas the height of the nanowires lies in the range of 8.5–14.0 nm from the substrate surface. The line profile on the top left insets of Fig. 4(b) has been measured on the nanoislands and that on the top right inset of Fig. 4(b) has been measured on the nanowires.

Let us try to understand the difference between the shapes of CoSi₂ and γ-FeSi₂ islands. CoSi₂ islands grow predominantly as equilateral triangular structures on the threefold symmetric Si(111) substrate. Earlier, gold-silicide islands were also found to grow on Si(111) as equilateral triangular structure up to a critical size; islands larger than the critical size undergoes a shape transition to trapezoidal islands [51]. We observe both equilateral and trapezoidal CoSi₂ islands in the present case. It is to be noted that shape transition of CoSi₂ islands on Si(100), which is fourfold symmetric, is from square-shaped islands to rectangular islands beyond the critical size; rectangular islands can grow into very long nanowires [33]. There are no such reported results for γ-FeSi₂ on Si(111) substrates. Here we observe that even the smallest γ -FeSi₂ islands appear to be elongated and the nanowires have grown along the three equivalent (110) directions. Logically, the longer edge of the trapezoidal CoSi2 islands also should be aligned along the three equivalent $\langle 1\,1\,0 \rangle$ directions. However, earlier studies of epitaxial growth on stepped Si(111) surfaces have shown that the trapezoidal islands

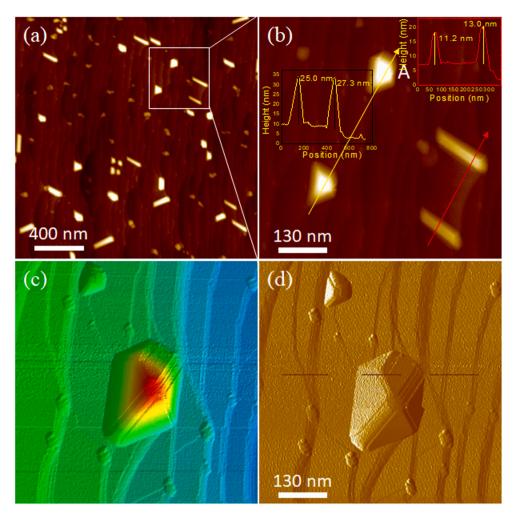


Fig. 4. (a-d) shows the STM images (bias voltage, $V_g=1.5~V$ and tunneling current, $I_t=0.2nA$) of FeSi $_2$ and CoSi $_2$ NWs and nanostructures on the Si(111)-7 \times 7 surface. Fig. 4 (b) shows a magnified image of the boxed part of (a). Fig. (c) shows the three-dimensional view of the nanoisland marked 'A' in the upper part in (b). (d) is the current image of its corresponding STM image, clearly displaying the distinct facets of the nanoisland.

preferentially grow along the step edges [50], as observed here. We believe that the difference in the observed shape of $CoSi_2$ and γ -FeSi_2 islands is related to the interface structure of these two systems. To our knowledge, no good quality interface structural analysis is available for γ -FeSi_2/Si(111). We can compare the cases of $CoSi_2/Si(110)$ [34] and γ -FeSi_2/Si(110) [35], for which high quality interface analyses are available. The $CoSi_2/Si(110)$ system is a case of simple epitaxy where the interface is simply A- or B-type, [34] as known from many other studies. However, for the γ -FeSi_2/Si(110) system, the interface structure is more complex. It is a case of tilted epitaxy and the resulting interface presents a network of intrinsic interfacial dislocations [35]; this can be understood in terms of generalized heteroepitaxy [52-54]. We believe, the difference of shapes between the $CoSi_2$ and γ -FeSi_2 nanostructures arises, at least partially, from their different interface structures.

3.3. X-ray photoelectron spectroscopy (XPS) measurements of the epitaxial FeSi $_2$ and CoSi $_2$ nanostructures on Si(111)-7 \times 7 surface

In Fig. 7, XPS core-level spectra of (a) Co 2p, (b) Fe 2p, (c) Si 2p, and (d) O 1 s, collected from the sample, depicting the co-existence of $CoSi_2$ and $FeSi_2$ formation on the Si(111) surface. Experimental core-level spectra (open circles) and fitted spectra (red) are plotted together and shifted along the *y*-axis for clarity. (a): Three pairs of components, used to fit the entire Co 2p core-level region, are also displayed. Features belonging to $CoSi_2$, $Co(OH)_2$ and satellites are shown by blue, wine, and

navy, respectively. (b): Three pairs of components used to fit the entire Fe 2p core-level region are displayed. Features belonging to FeSi₂, FeO and satellites are shown by blue, dark yellow, and navy, respectively. (c): Two pairs of components used to fit the entire Si 2p core-level region are displayed. Features belonging to Si(111) and (Fe, Co)Si₂ are shown by wine and blue, respectively. (d): Two components used to fit the O 1 s core-level region are displayed. Features belonging to ${\rm O}^{2-}$ and ${\rm OH}^-$ are shown by blue and pink, respectively.

We have investigated the Co 2p, Fe 2p, Si 2p and O 1 s core-level spectra, which are shown in Fig. 7(a), 7(b), 8(c) and 7(d), respectively. The Co $2p_{3/2}$ and Co $2p_{1/2}$ spin–orbit doublet peaks are centered at $\sim 778.6~\text{eV}$ and $\sim 793.6~\text{eV}$, respectively [55]. The spin–orbit doublet peaks are wide and show two components in each doublet, a lower binding energy pair of peaks which are related to the CoSi $_2$ components and other higher binding energy pair of peaks are related to Co(OH) $_2$ components. We have carefully fitted the spectrum using following fitting parameters for CoSi $_2$ components as shown by blue line shapes: spin–orbit splitting 15.0 eV, branching ratio 2.5, full widths at half maximum (FWHM) 2.6 eV and an integral background being subtracted before fitting. While Co(OH) $_2$ components are shown by wine line shapes obtained by using spin–orbit splitting 15.3 eV, branching ratio 5.0 and FWHM 3.8 eV. The satellite peaks are observed at $\sim 7.0~\text{eV}$ above for CoSi $_2$ shown by navy line shapes [55].

The Fe $2p_{3/2}$ and Fe $2p_{1/2}$ spin-orbit doublet peaks are located at around 707 eV and 719.9 eV, respectively with a pair of shake-up

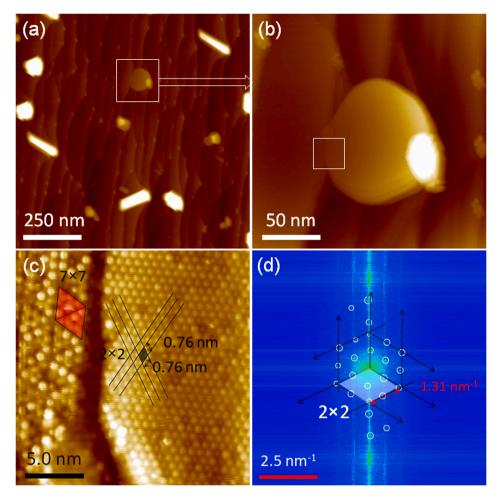


Fig. 5. The STM image shows a flat terrace silicide island in (a-c). (c) shows the 2×2 - R30° reconstructed atomically resolved STM image of the top surface of the silicide island of the box marked location in (b). (d) is the Fast Fourier transform (FFT) image of (c).

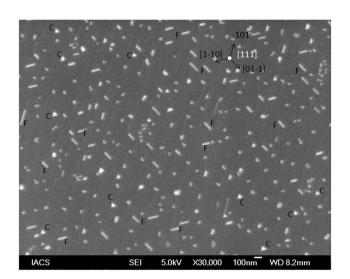


Fig. 6. The SEM image shows the co-existence of FeSi $_2$ nanowires and CoSi $_2$ triangular islands on Si(111) surfaces. The Nanowires grow predominantly in three crystallographically symmetric in-plane $\langle 110 \rangle$ directions.

satellite peaks located at 8.8 eV above of their spin–orbit doublet peaks [56]. The fitted Fe 2p spectrum highlights characteristic doublet peaks of FeSi2 species at ~ 707.0 eV and ~ 719.8 eV for Fe 2p $_{3/2}$ and Fe 2p $_{1/2}$, respectively, and doublet peaks of FeO species at ~ 708.3 eV and \sim

721.8 eV for Fe $2p_{3/2}$ and Fe $2p_{1/2}$, respectively. This suggests the presence of FeSi₂ and iron(II)-oxide (FeO) at the surface of Si(111). The presence of Si(111), FeSi₂ and CoSi₂ were visible from the characteristics peaks of Si 2p core-level spectrum [56]. The Si $2p_{3/2}$ and Si $2p_{1/2}$ spin–orbit doublet peaks are located at around 99.2 eV and 99.8 eV, respectively corresponding to Si(111) species. The higher binding shoulders at around 99.4 eV and 100.0 eV for Si $2p_{3/2}$ and Si $2p_{1/2}$, respectively correspond to FeSi₂ and CoSi₂ species. The presence of O²⁻ and OH⁻ peaks at \sim 531.6 eV and \sim 532.8 eV, respectively in the O 1s core-level further demonstrates the formation of silicide, oxides/hydroxyl species of cobalt and iron on the surface of Si(111) as shown in Fig. 7(d). Thus, the spectroscopic investigation showed the formation of cobalt and iron disilicide on Si(111) [56].

4. Summary and conclusions

Self-organized epitaxial FeSi $_2$ NWs and CoSi $_2$ nano-islands have been grown on the same Si(111)-7 \times 7 surface for the first time. The morphology has been investigated by in-situ scanning tunneling microscopy and field emission scanning electron microscopy. The spectroscopic characteristics of FeSi $_2$ and CoSi $_2$ have been confirmed by x-ray photoelectron spectroscopy measurements. FeSi $_2$ grows as nanowires and CoSi $_2$ grows predominantly as nanoislands. This investigation establishes that CoSi $_2$ nanostructure grows on the exposed Si(111)-7 \times 7 surface of the FeSi $_2$ -decorated substrate upon Co deposition. FeSi $_2$ nanostructures remain stable while heating the sample at \sim 600 °C for cobalt deposition. There is no noticeable influence of the pre-existing FeSi $_2$ nanostructures on the Si(111)-7 \times 7 surface. The two types of

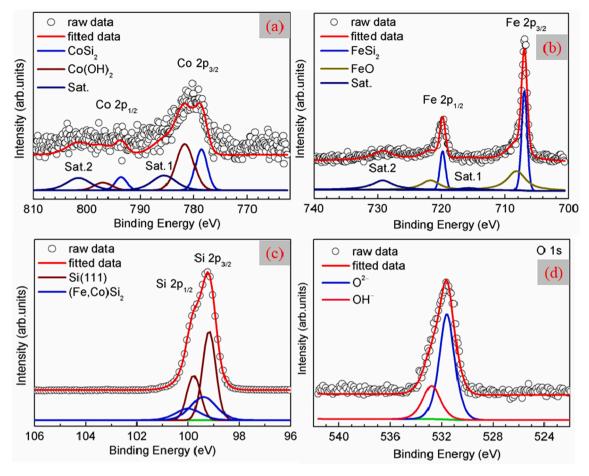


Fig. 7. The XPS data confirms the co-existence of FeSi2 and CoSi2 islands on the Si(111) surface.

di-silicides fabricated on the same $Si(111)-7 \times 7$ surface may be utilized in the nanodevice fabrication. For ferromagnet-superconductor hybrid structure applications in quantum technology, it is important to control the separation between the ferromagnetic and the superconducting, in other words between FeSi2 and CoSi2, nanostructures on the substrate. This separation can be controlled via the amount of Fe and Co deposition.

CRediT authorship contribution statement

J.C. Mahato: Conceptualization, Methodology, Visualization, Investigation, Formal analysis, Writing - original draft. Debolina Das: Investigation, Formal analysis. Arindam Pal: Investigation, Formal analysis. Prabir Pal: Investigation, Formal analysis. B.N. Dev: Funding acquisition, Resources, Visualization, Supervision, Writing - review & editing.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Association for the Cultivation of Science, Kolkata, India.

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